

Magnetic Nanoparticle Formation From Ferric Acetylacetonate Decomposition on Cold Substrates using Supercritical Fluids

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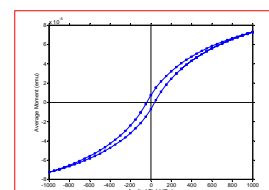
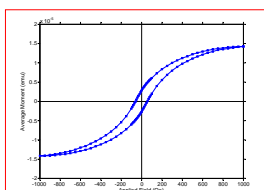
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Introduction and Overview

New synthetic routes for the preparation of magnetic nanoparticles and thin films are under constant investigation. In particular iron oxide magnetic particles have attracted interest due to their application as recording media, ferrofluids, catalysts and targeted drug delivery. Ferric acetylacetonate $Fe(acac)_3$ is known to undergo thermal decomposition to form either Fe_3O_4 or $\alpha-Fe_2O_3$ magnetically ordered materials when heated above 180 °C. We decomposed a mixture of ferric acetylacetonate ($Fe(acac)_3$) in supercritical CO_2 to deposit magnetic nanoparticle films at room temperature. Two different processes were developed: one based on the Rapid Expansion of a Supercritical Solution (RESS) and another based on the depressurization of the saturated supercritical mixture in a fixed volume (BATCH process). The rate of decomposition varies dramatically between the two processes, RESS (μ sec scale) and BATCH (sec scale).

Results: Magnetic Measurements

Quantum Design SQUID magnetometer



- ❖ Ferromagnetic signal with similar properties at 2K and 300K for both RESS and BATCH processes
- ❖ Comparable paramagnetic signal strengths only for RESS samples
- ❖ At 300K: $M_s = 7.3 \times 10^{-5}$ emu, $H_c = 35$ Oe, $M_r = 2.1 \times 10^{-6}$ emu (BATCH)
- ❖ At 300K: $M_s = 1.4 \times 10^{-5}$ emu, $H_c = 55$ Oe, $M_r = 8.3 \times 10^{-6}$ emu (RESS)
- ❖ $Fe(acac)_3$ starting material is paramagnetic

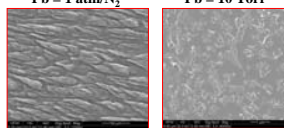
$Fe(acac)_3$ decomposition to occur at $T < T_{decomp}$!

Results: SEM imaging

RESS process

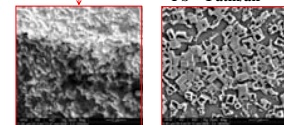
Pb = 1 atm/ N_2

Pb = 10 Torr



Size: 30-100 nm

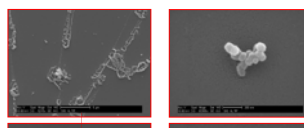
Pb = 1 atm/air



Size: 100-700 nm

Size: 80-600 nm

BATCH process



Slow Depressurization

Fast Depressurization

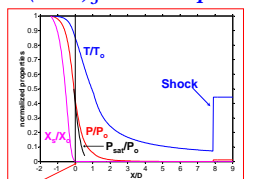
Size: 10-100 nm

$P_0 = 140$ bar, $T_0 = 313$ K
 $X_{Fe} = 1.4 \times 10^{-6}$ mole fraction

Smaller nanoparticles are formed when the expansion is into vacuum

Effect of shockwave structure?

Calculated Centerline Properties of $CO_2/Fe(acac)_3$ RESS Expansion

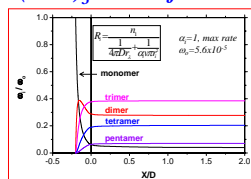


Nozzle exit

$P_0 = 140$ bar, $T_0 = 343$ K, $D = 100$ μ m

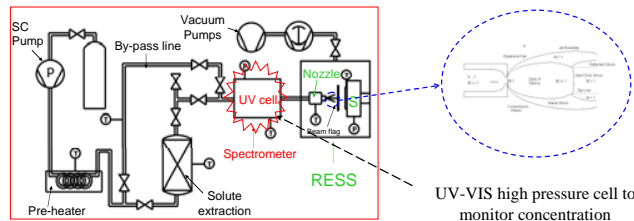
Since the RESS expansion is predicted to provide less than 10 nm scale clusters, substantial growth must occur at the surface

Diffusion-kinetic model of $Fe(acac)_3$ cluster formation



RESS Experimental Apparatus

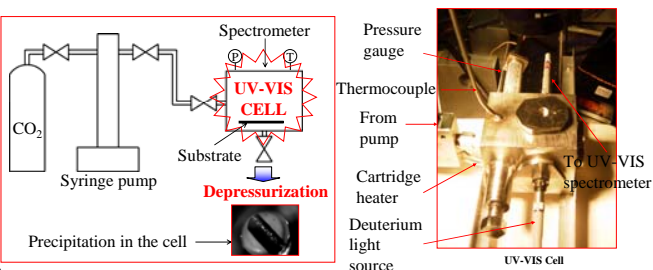
The magnetic thin films are produced by expanding the supercritical solution and directing the resulting supersonic jet onto cold silicon wafers



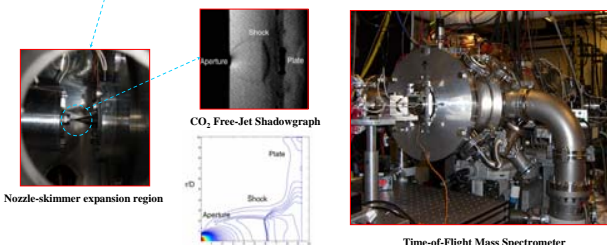
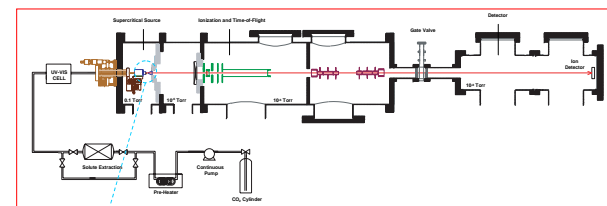
UV-VIS high pressure cell to monitor concentration

BATCH Experimental Apparatus

Depressurization of the supercritical mixture within the fixed volume UV-VIS cell causes the solute to precipitate and deposit on a silicon wafer, placed within the cell.

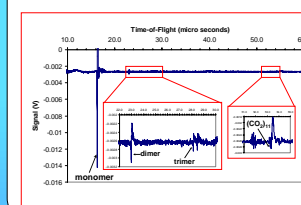


Interface of RESS Source with a TOF Mass Spectrometer



- ❖ Sample a molecular beam from the high pressure supersonic free-jet RESS expansion
- ❖ Identify size and composition of the clusters and nanoparticles formed in the jet
- ❖ Compare experimental data with kinetic models

Preliminary Results



- ❖ Verified the detection of CO_2 monomer and CO_2 clusters in the RESS expansion
- ❖ Need to optimize the interface between the RESS expansion and the Time-of-Flight Mass Spectrometer
- ❖ Improve detection of higher order clusters

Conclusions

- ❖ Thin magnetic nanoparticle films were produced on cold substrates with RESS and BATCH processes
- ❖ All films were magnetically ordered with coercivities between 30-60 Oe
- ❖ Heating of $Fe(acac)_3$ above T_{decomp} to produce the magnetic nanoparticles was not necessary
- ❖ The mechanism for $Fe(acac)_3$ decomposition in supercritical CO_2 at $T < T_{decomp}$ needs further investigation
- ❖ The RESS expansion was coupled with a Time-of-Flight Mass Spectrometer and preliminary measurements were performed

Acknowledgements

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